Coral Clinical Systems
A Division of Tulip Diagnostics (P) Ltd.
Verna, Goa – 403 722, INDIA

Safety Data Sheet

1. INFORMATION ON THE SUBSTANCE / PREPARATION AND COMPANY

1.1 Product Name : **Elyte 3 Kit (NA^+, K^+ & Cl^- Colorimetric Method)**

Kit	Catalog No.	Components
Elyte 3 Kit (NA^+, K^+ & Cl^- Colorimetric Method)	1103060015	**Sodium Kit** L1: Precipitating Reagent L2: Acid Reagent L3: Colour Reagent **Potassium Kit** L1: Potassium Reagent S: Na^+ & K^+Standard **Chloride Kit** CL1: Chloride Reagent CS: Chloride Standard

1.2 Intended Use : For the determination of NA^+, K^+ & Cl^- in serum.

1.3 Company : **Coral Clinical Systems**
(A Division of Tulip Diagnostics (Pvt) Ltd.)
Plot No. M-46 & M-47, Building E,
Phase III B, Verna Industrial Estate,
Verna-Goa, 403722
Tel :- +91-832-6680121
Fax :- +91-832-2887028
E-mail :- coral@tulipgroup.com

1.4 In emergencies :Call your local emergency center.

2. Composition / Information on Hazardous Ingredients

Chemical Name	CAS #	% W/V	Exposure Limits in Air				
			ACGIH		OSHA		OTHER
			TLV	STEL	PEL	STEL	
Reagent 1: Chloride Reagent							
Ferric Nitrate	7782-61-8	1%	NA	NA	NA	NA	NA

3.Hazard Identification

According to 1999/45/EC, the preparation is classified as hazardous substance or mixture.
Skin Corrosion,category 1B, H314

Other Information

The reagents do not contain materials from animal origin

4.First Aid Measures

Inhalation:

If breathing becomes difficult, remove victim to fresh air. Seek medical attention.

Ingestion:

Get medical attention if there has been ingestion of this product.

Skin Contact:

Immediately flush skin with plenty of water for atleast 15 minutes. Remove contaminated cloths and shoes. Get medical attention immediately.

Eye Contact:

Immediately flush eye(s) with large volume of water for atleast 15 minutes, occasionally lifting the lower lids. Get medical attention immediately.

5.Fire Fighting Measures

Flash Point (Method used): NA **Flammable Limits – LEL:** NA **UEL:** NA

Extinguishing Media:

Use fire-extinguishing media appropriate for site conditions.

Special Fire Fighting Procedures:

Structural firefighting gear and self-contained breathing apparatus will provide adequate protection if this product is in a fire area.

Unusual Fire and Explosion Hazards:

Ethanol: highly flammable and is likely to catch fire; Ferric Nitrate: promotes fire started and impedes fire fighting.

6.Accidental Release Measures

Steps to be taken in case material is Released or Spilled:

Use an absorbent material to contain / pick up the spilled solution. Place all spill residues into a suitable container, seal, label and hold for disposal.

7.Handling and Storage

Refer to packet insert for additional information on handling and storage procedures.

8.Exposure Controls and Personal Protection

Ventilation Data:

A system of local and / or general exhaust is recommended to keep employee exposures as low as possible. Local exhaust ventilation is generally preferred because it can control the emissions of the contaminant at its source.

Respiratory Protection:

Respiratory protection is not required under normal use of this product. If respiratory protection is needed, follow OSHA respirator regulations (29CFR1910.134) and, if necessary, wear a NIOSH approved respirator. Select respirator based on its suitability to provide worker protection for given working conditions, level of airborne concentration, and presence of sufficient oxygen.

Protective Gloves:

Wear appropriate protective gloves to prevent skin contact.

Other Protective Equipment:
Wear appropriate eye protection to prevent eye contact. Wear appropriate body protection to prevent skin contact.

Other Engineering Controls:
Eye wash stations and deluge showers.

Work Practices:
Good laboratory technique should be used when handling this product. Observe appropriate chemical hygiene. Avoid contact with eyes or skin. Do not place in mouth.

Hygienic Practices:
Do not eat, drink, or smoke while working with product. Upon completion of work activities involving this product, wash hands thoroughly with soap and water.

9. Physical And Chemical Properties

L1: Precipitating Reagent

Relative Vapour density(air = 1)	:	NA	Evaporation rate(nBuAc = 1):		NA
Specific Gravity (water = 1)	:	NA	Freezing / Melting Point	:	NA
Solubility in Water	:	Soluble	Boiling Point	:	NA
Vapour Pressure, mm Hg @ 20°C:		NA	pH	:	NA

L2: Acid Reagent

Relative Vapour density(air = 1)	:	NA	Evaporation rate(nBuAc = 1):		NA
Specific Gravity (water = 1)	:	NA	Freezing / Melting Point	:	NA
Solubility in Water	:	Soluble	Boiling Point	:	NA
Vapour Pressure, mm Hg @ 20°C:		NA	pH	:	NA

L3: Colour Reagent

Relative Vapour density(air = 1)	:	NA	Evaporation rate(nBuAc = 1):		NA
Specific Gravity (water = 1)	:	NA	Freezing / Melting Point	:	NA
Solubility in Water	:	Soluble	Boiling Point	:	NA
Vapour Pressure, mm Hg @ 20°C:		NA	pH	:	NA

L1: Potassium Reagent

Relative Vapour density(air = 1)	:	NA	Evaporation rate(nBuAc = 1):		NA
Specific Gravity (water = 1)	:	NA	Freezing / Melting Point	:	NA
Solubility in Water	:	Soluble	Boiling Point	:	NA
Vapour Pressure, mm Hg @ 20°C:		NA	pH	:	NA

S: Na^+ & K^+ Standard

Relative Vapour density(air = 1)	:	NA	Evaporation rate(nBuAc = 1):		NA
Specific Gravity (water = 1)	:	NA	Freezing / Melting Point	:	NA
Solubility in Water	:	Soluble	Boiling Point	:	NA
Vapour Pressure, mm Hg @ 20°C:		NA	pH	:	NA

CL1: Chloride Reagent

Relative Vapour density(air = 1)	:	NA	Evaporation rate(nBuAc = 1):		NA
Specific Gravity (water = 1)	:	NA	Freezing / Melting Point	:	NA
Solubility in Water	:	Soluble	Boiling Point	:	NA
Vapour Pressure, mm Hg @ 20°C:		NA	pH	:	1.4±0.1

CS: Chloride Standard

Relative Vapour density(air = 1)	:	NA	Evaporation rate(nBuAc = 1):		NA
Specific Gravity (water = 1)	:	NA	Freezing / Melting Point	:	NA
Solubility in Water	:	Soluble	Boiling Point	:	NA
Vapour Pressure, mm Hg @ 20°C:		NA	pH	:	NA

Odour and Appearance Information

Reagent 1: Clear, Yellowish solution

Reagent 2: Clear, colourless solution

Reagent 3: Clear, Yellowish solution

Reagent 4: Clear to slightly hazy solution

Reagent 5: Clear ,colourless solution

Reagent 6: Clear, pale yellow solution

Reagent 7: Clear ,colourless solution

10.Stability and Reactivity

Incompatibility (Materials to Avoid):
Strong oxidizer. Keep away from all flammable substances.

Hazardous Decomposition Products:
NA.

Will Hazardous Polymerization Occur?
NA.

Conditions to Avoid / Polymerization:NA.

Is the Product Stable?
Yes, under normal handling and storage conditions.

Conditions to Avoid/stability
Stable Solution. Avoid from naked flame or sparks

11.Toxicological Information

Toxicity Data:
Ferric Nitrate: Oral rat LD50: 3250mg/kg.

Reproductive effects:
NA.

Target organ Effects:
Eyes, skin, Central nervous systems, & cardiovascular systems.

Carcinogenicity: No

CHEMICAL NAME	CAS #	% W/V	NTP Carcinogen		IARC	OSHA
			Known	Anticipated		
NA.						

12.Ecological Information

Environmental Fate / Stability:NA

Effect of Material on plants or animals:NA

Effect of Chemical on Aquatic Life:NA

13.Disposal Considerations

EPA Waste Number and Proper Waste Disposal Method:
Please consult local, state and federal regulations for additional guidance on disposal.

14.Transportation Information

Is this Material Hazardous? Not regulated under transportation regulations.

Proper Shipping Name : NA		
Hazard Class Number : NA	Packing Group: NA	UN Number: NA

15.Regulatory Information NA.

16.Other Information

Hazard Statement and Precautionary statements referred elsewhere in the safety data sheet:
H314-Causes severe skin burns and eye damage.

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